

Rong Wan,^{a*} Feng Wu,^a Lin Cao^b
and Jin-Tang Wang^a^aDepartment of Applied Chemistry, College of Science, Nanjing University of Technology, No. 5 Ximofan Road, Nanjing 210009, People's Republic of China, and ^bDepartment of Pharmaceutical Analysis, China Pharmaceutical University, No. 24 Tongji Xiang, Nanjing 210009, People's Republic of China

Correspondence e-mail: rwan01@jlonline.com

Key indicators

Single-crystal X-ray study

T = 293 K

Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$

R factor = 0.045

wR factor = 0.146

Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

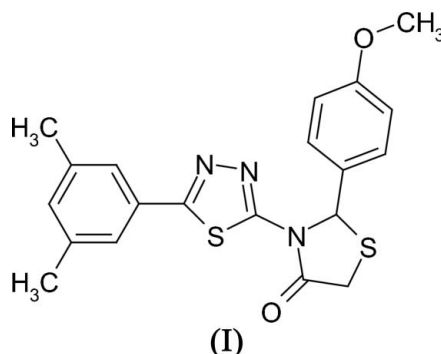
3-[5-(3,5-Dimethylphenyl)-1,3,4-thiadiazol-2-yl]-2-(4-methoxyphenyl)thiazolidin-4-one

The title compound, $\text{C}_{20}\text{H}_{19}\text{N}_3\text{O}_2\text{S}_2$, crystallizes with $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds which form a three-dimensional network.

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Comment

Thiadiazole derivatives containing the thiazolidinone unit are of great interest because of their chemical and pharmaceutical properties. Some derivatives have fungicidal and herbicidal activities (Arun *et al.*, 1999) and others show insecticidal activities (Wasfy *et al.*, 1996). We report here the crystal structure of the title compound, (I).The molecular structure of (I) is shown in Fig. 1. Intermolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds form a three-dimensional network (Table 1 and Fig. 2).

Experimental

For the preparation of the title compound, [5-(3,5-dimethylphenyl)-1,3,4-thiadiazol-2-yl]-(4-methoxybenzylidene)amine (5 mmol) and mercaptoacetic acid (5 mmol) were added to toluene (50 ml). The water produced by the condensation reaction was removed by distillation over a period of 5 h. The reaction mixture was then left to cool to room temperature and filtered; the filter cake was recrystallized from acetone to give pure compound (I) (m.p. 463–464 K). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an acetone solution.

Crystal data

 $\text{C}_{20}\text{H}_{19}\text{N}_3\text{O}_2\text{S}_2$ $M_r = 397.50$ Monoclinic, $P2_1/c$ $a = 13.892 (3) \text{ \AA}$ $b = 7.2560 (15) \text{ \AA}$ $c = 18.912 (4) \text{ \AA}$ $\beta = 95.86 (3)^\circ$ $V = 1896.4 (7) \text{ \AA}^3$

Z = 4

 $D_x = 1.392 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation $\mu = 0.30 \text{ mm}^{-1}$

T = 293 (2) K

Block, colourless

0.30 × 0.20 × 0.10 mm

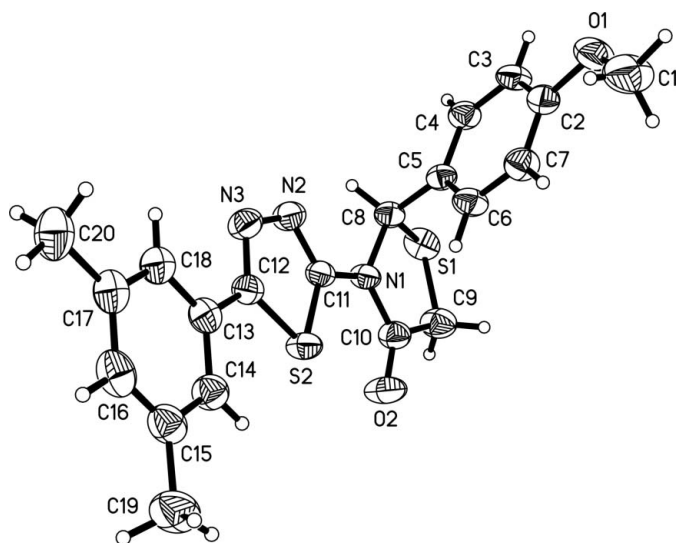


Figure 1
A view of the molecular structure of (I). Displacement ellipsoids are drawn at the 50% probability level.

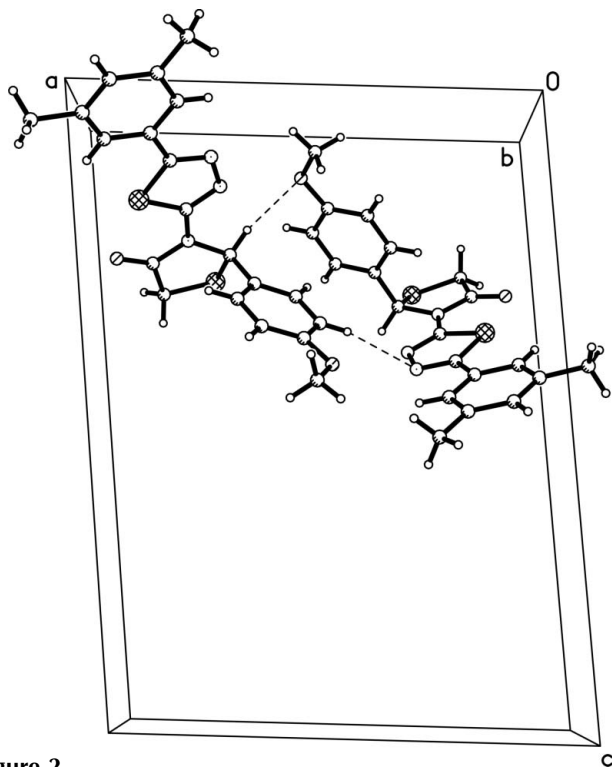


Figure 2
The hydrogen bonding in (I), with dashed lines indicating intermolecular C—H...O and C—H...N hydrogen bonds.

Data collection

Enraf–Nonius CAD-4
diffractometer
 $\omega/2\theta$ scans
Absorption correction: ψ scan
(North *et al.*, 1968)
 $T_{\min} = 0.915$, $T_{\max} = 0.971$
3866 measured reflections

3709 independent reflections
2904 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.019$
 $\theta_{\max} = 26.0^\circ$
3 standard reflections
every 200 reflections
intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.146$
 $S = 1.03$
3709 reflections
244 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.36 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.29 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C3-H3B\cdots N3^i$	0.93	2.62	3.533 (3)	169
$C8-H8A\cdots O1^i$	0.98	2.57	3.530 (3)	166

Symmetry code: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$.

All H atoms were positioned geometrically, with C—H distances in the range 0.93–0.97 \AA , and included in the refinement in a riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5U_{\text{eq}}(\text{C})$

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

References

- Arun, K. P., Nag, V. L. & Panda, C. S. (1999). *Indian J. Chem. Sect B*, **38**, 998–1001.
- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
- Siemens (1996). *SHELXTL*. Version 5.06. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Wasfy, A. A., Nassar, S. A. & Eissa, A. M. (1996). *Indian J. Chem. Sect B*, **35**, 1218–1220.